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The Carbon Cycle

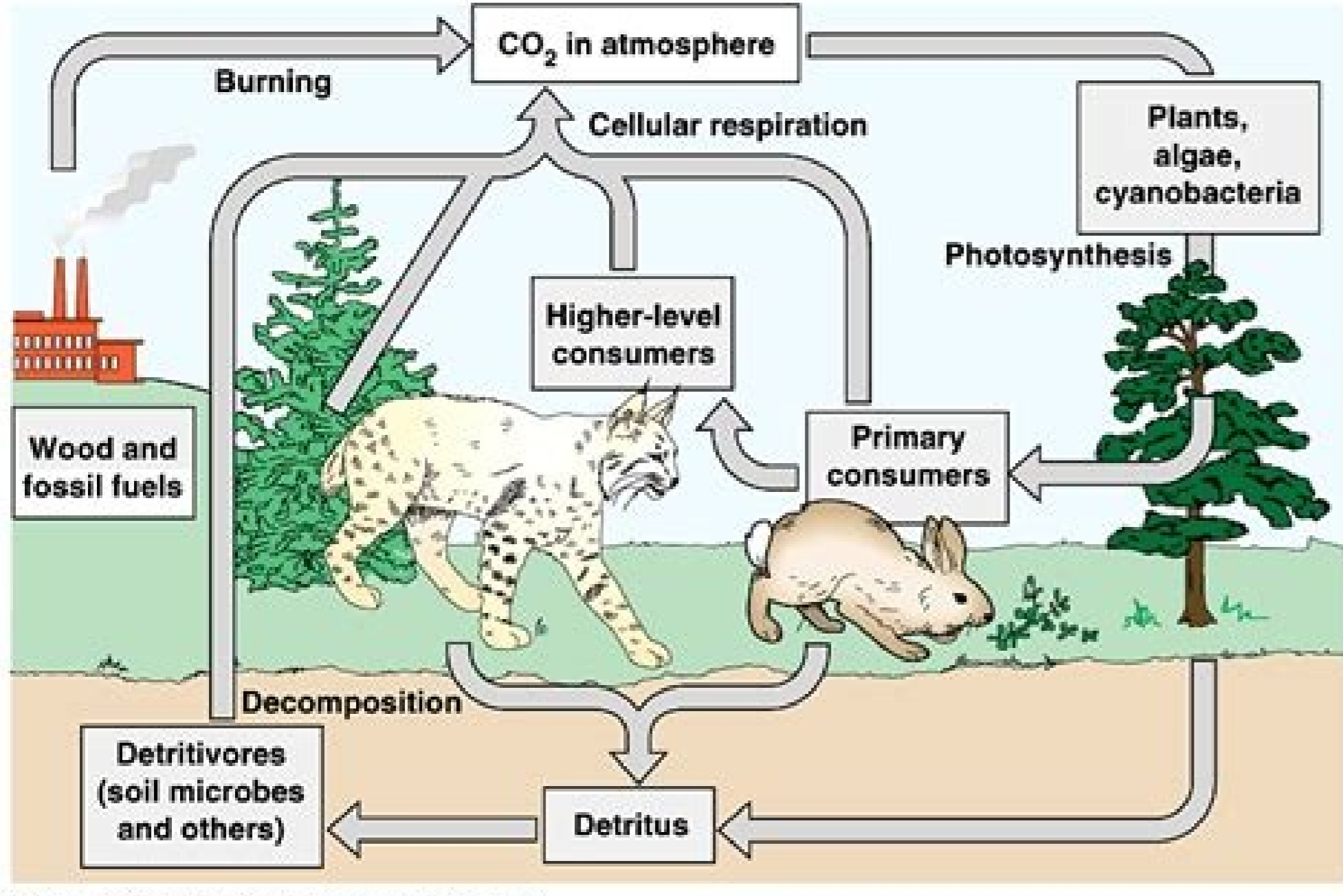
1. How is carbon used by living things? (2)
2. How is carbon stored in the ground? (2)
3. How is carbon stored in the atmosphere? (2)
4. How is carbon stored in the oceans? (2)
5. How is carbon stored in the rocks? (2)

The Carbon Cycle

Use the carbon cycle diagram to help you answer the questions.

1. How is carbon used by living things? (2)

ink saving Eco



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Cycling of Matter in Ecosystems

- The Water Cycle
- The Carbon Cycle
- The Nitrogen Cycle
- The Phosphorus Cycle

Nitrogen Cycle

1. The process by which nitrogen is converted from an inorganic form into an organic form by plants and animals is called **nitrogen fixation**.

2. The process by which nitrogen is converted from an organic form into an inorganic form by decomposers is called **denitrification**.

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4. The process by which nitrogen is converted from an organic form into an inorganic form by decomposers is called **denitrification**.

The Carbon Cycle

Name: _____ Date: _____

1. Label the steps of the carbon cycle using the diagram above.
 1. _____
 2. _____
 3. _____
 4. _____
 5. _____
 6. _____
 7. _____
2. During _____ plants take in carbon dioxide, and during _____ humans release carbon dioxide.
3. What role do decomposers play in the carbon cycle? _____
4. What impact do humans have on the carbon cycle? _____
5. Name 2 carbon sinks: _____ and _____

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Why are nitrogen and phosphorus both in group 5. Crash course nitrogen and phosphorus cycle guided viewing worksheet answers. Crash course nitrogen and phosphorus cycle worksheet answers.

A schematic of the modelled nitrogen and phosphorus cycles. The unlabelled arrows leading into the phytoplankton (O and NF) indicate the biological uptake of nitrate and phosphate. The article provides answers to the questions posed in the nitrogen-phosphorus model worksheet. Worksheet answers 1. Try some different types of initial conditions: e.g. (a) low nitrate (N), all else the same; (b) high N; (c) low P; (d) high P. Does the model always converge to a steady state? Yes. Starting, for instance, from deep [NO₃] = 2.3 or 50.3, or deep [PO₄] = 0.2 or 3.0, the model always converges towards a stable final state (flat lines in all graphs). 2. Is it always the same steady state? Yes. N₂-fixers = 0.02 μMol P m⁻³; non-fixers = 2.72 μMol P m⁻³; surface [PO₄] = 0.15 mMol m⁻³; deep [PO₄] = 1.75 mMol m⁻³; surface [NO₃] = 2.0 mMol m⁻³; deep [NO₃] = 25.6 mMol m⁻³. 3. Try starting the model at steady state except for ten-fold less PO₄ in the deep box. Try the same run in the P-only model. Is the behaviour the same in both models? In terms of surface and deep P concentrations, do the two types of model converge to the same sorts of P concentrations? P behaviour is near-identical in the two models (slight differences, but not important). 4. Run the model in steady-state and examine the fluxes. Now run the model again and examine the fluxes shortly after: (a) doubling the deep nitrate concentration, and (b) after halving it. Which fluxes have changed between the three runs? Are the changes in the right direction to restore the N:P ratio to its steady-state value? (to calculate this, use the facility to "save data" and then examine it in Excel or Matlab) They do agree. Only the N₂-fixation flux is different. Yes, the changes act to return the N:P ratio to its steady-state value (negative feedbacks). 5. Try quadrupling the denitrification rate (the fraction of organic biomass that is remineralised via denitrification) in the standard model and examine the new steady-state that the model converges to. The denitrification flux should now equal -488 Tg N y⁻¹. Assuming other fluxes are as in the first figure overlaid, how large would the N₂-fixation flux need to be in order to create a balance between inputs and outputs? Is it that size? Without N₂-fixation the imbalance is 488 + 16 - 30 - 38 = 436 Tg N y⁻¹. N₂-fixation takes up the slack (increases to that value, in steady-state). Nitrogen-fixation increases because denitrification removes reactive nitrogen (represented by nitrate in the model) from the system, and so conditions are made more favourable for those phytoplankton able to get nitrogen from other sources, i.e. N₂ (which is always plentiful). 6. Similarly with the atmospheric input of nitrogen, what effect does a 25% reduction in AN (atmospheric nitrogen deposition) have on steady-state fluxes? Again nitrogen-fixation increases to take up the slack and bring extra nitrogen into the system. 7. From the steady-state, try increasing RP (river delivery of phosphate) by 25%. What effect does this have on TPP? Reset the model parameters to their default values, run to steady-state again, then try increasing RN (river delivery of nitrate) by 25%. What effect does this have on TPP? Increasing RP increases TPP proportionately. Increasing RN has no effect. 8. Look at the figures from the increased RN run. Why does increasing RP have an effect on TPP, but increasing RN have none? How is this possible when surface nitrate is more limiting to the "other" phytoplankton? Increased RP - more PO₄ in the system - increased nitrogen fixation (because extra PO₄ not initially accompanied by extra NO₃) - more NO₃ brought into the system - more of both PO₄ and NO₃ - increased TPP. Conversely, increased RN - more NO₃ in the system - decreased nitrogen fixation (because extra NO₃ not initially accompanied by extra PO₄) - less NO₃ brought into the system - no net increase in either PO₄ or NO₃ - no effect on TPP. 9. Can you explain how/why it is possible to have distinct proximate and ultimate limiting nutrients? It is because the population of nitrogen-fixers is limited and controlled by phosphate: in the model, nitrogen-fixer growth rate ($\mu = \mu_{max} * PLIM$) is dependent only on phosphate. Although nitrogen-fixers typically make up only a small fraction of the total phytoplankton community, they also act as a conduit for bring new nitrate into the system and in this way their activities also influence the nutrient levels for the other phytoplankton.

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